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TESTING LABORATORY

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An Introduction to Isotopic Tracers in Stormwater Runoff



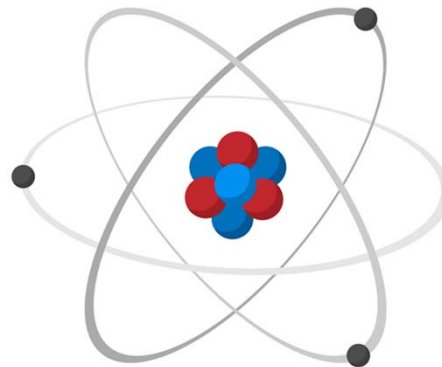
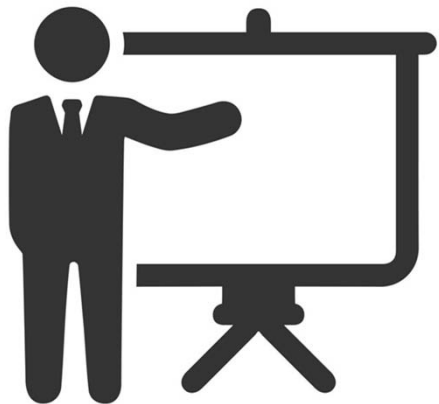
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Sean P. Ahearn, MSc

The Goals

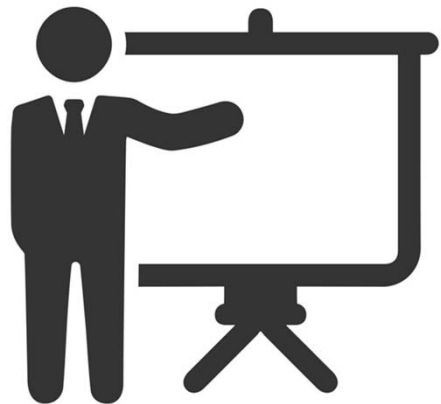
- ❑ Define isotopic fingerprinting of water and nutrients.
- ❑ Explain how fingerprinting is used to ID and track water and nutrient sources.
- ❑ Review case study applications.



Our Road Map

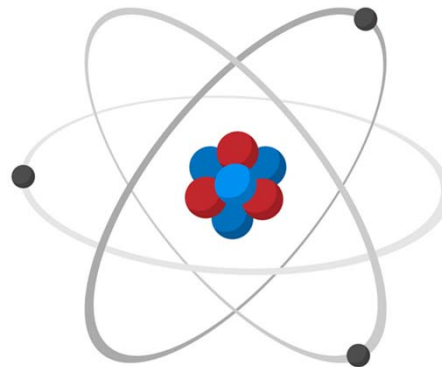
Introduction & Background

- ❑ Beta Analytic
 - ❑ What is an isotope?
 - ❑ What is a delta value?
 - ❑ Fingerprinting
 - ❑ The Water Cycle



Nutrient Source Tracking

- ❑ Why Track Nutrients?
 - ❑ Nitrate (NO_3^-) $\delta^{18}\text{O}$ & $\delta^{15}\text{N}$
 - ❑ Stormwater Applications
 - ❑ Review
 - ❑ Questions?





Beta Analytic
TESTING LABORATORY

- The **ISO/IEC 17025:2005 accreditation** is the highest level of recognized quality any testing or calibration laboratory can attain.



BETA is celebrating **40** years of testing

- **Radiocarbon age/activity** in archaeological, geological, and water samples
- **Stable isotope ratios** of carbon, deuterium, nitrogen, and oxygen in organic and carbonate materials and water via IRMS/CRDS

BETA is the certifying body for bio-based testing for the following



$\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ Reveal the Sources of Nitrate-Nitrogen in Urban Residential Stormwater Runoff

Yun-Ya Yang and Gurpal S. Toor*

Soil and Water Quality Laboratory, Gulf Coast Research and Education Center, University of Florida, Institute of Food and Agricultural Sciences, 14625 CR 672, Wimauma, Florida 33598, United States

Purpose:

-This study combined dual isotope source identification techniques with chemical analysis to interpret the transport and sources of NO_3^- -N from a low-density residential catchment.

Evaluation:

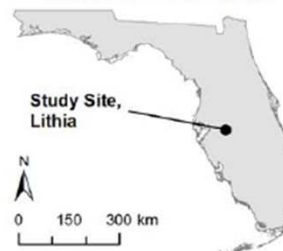
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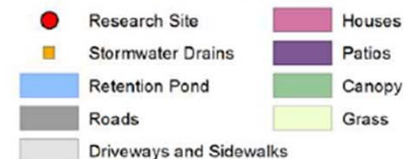
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State of Florida



Map Legend



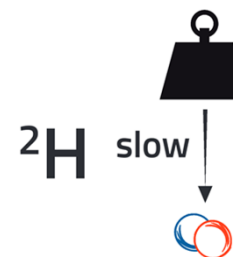
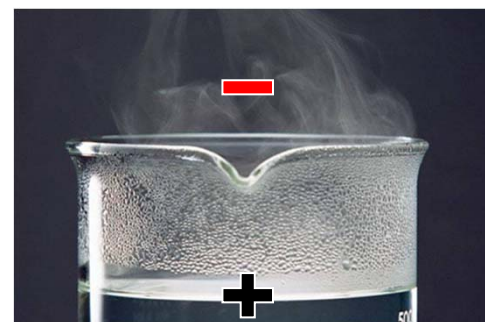
Stable Isotopes Review

-**Stable isotopes** are variations of the same element but with differing number of neutrons and therefore mass. This changes the mass of the element without changing the element's identity or overall reactivity.

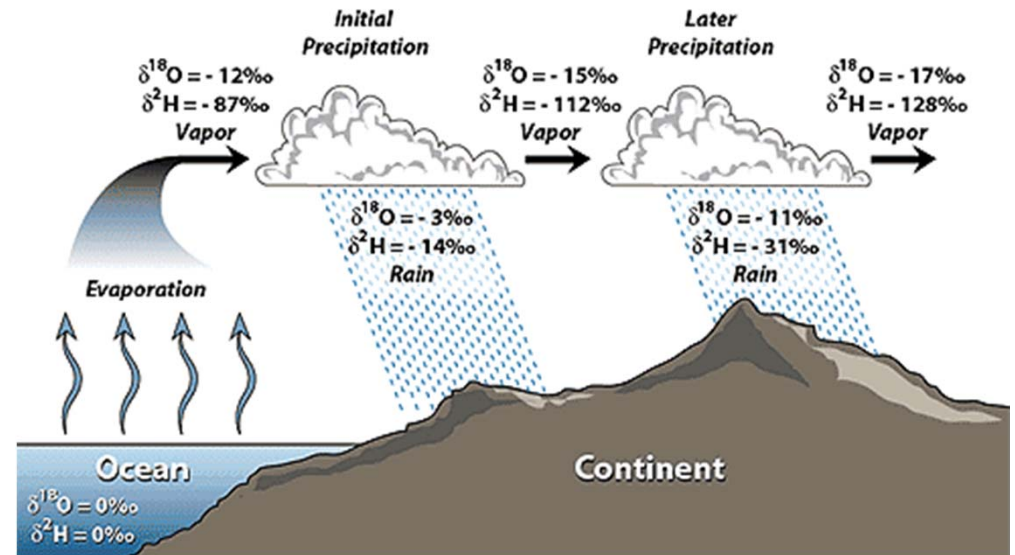
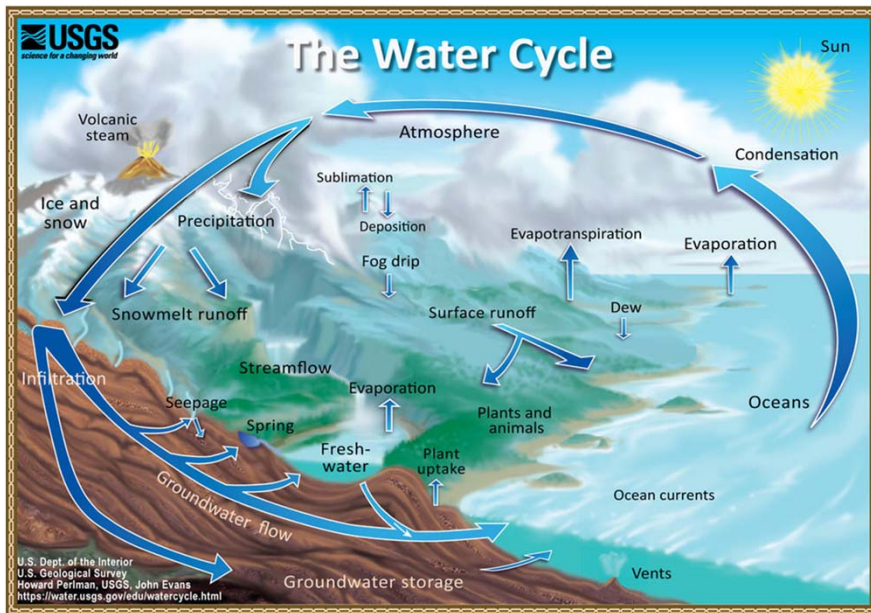
-Changes in delta values suggest that a reaction or phase change has taken place, fractionating the atom pool, expressed as the delta value and referred to as **depleted** and **enriched**.

^1H 1.00794 99.985% Stable	^2H 2.0141 0.015% Stable
--	--

$$\delta ^2H = \left(\frac{\left(\frac{^2H}{^1H} \right)_{Unknown}}{\left(\frac{^2H}{^1H} \right)_{Standard}} - 1 \right) * 1000 \text{ ‰}$$



Fingerprinting & The Water Cycle



<http://web.sahra.arizona.edu/programs/isotopes/oxygen.html>

-Natural processes imprint on the O and H delta values.

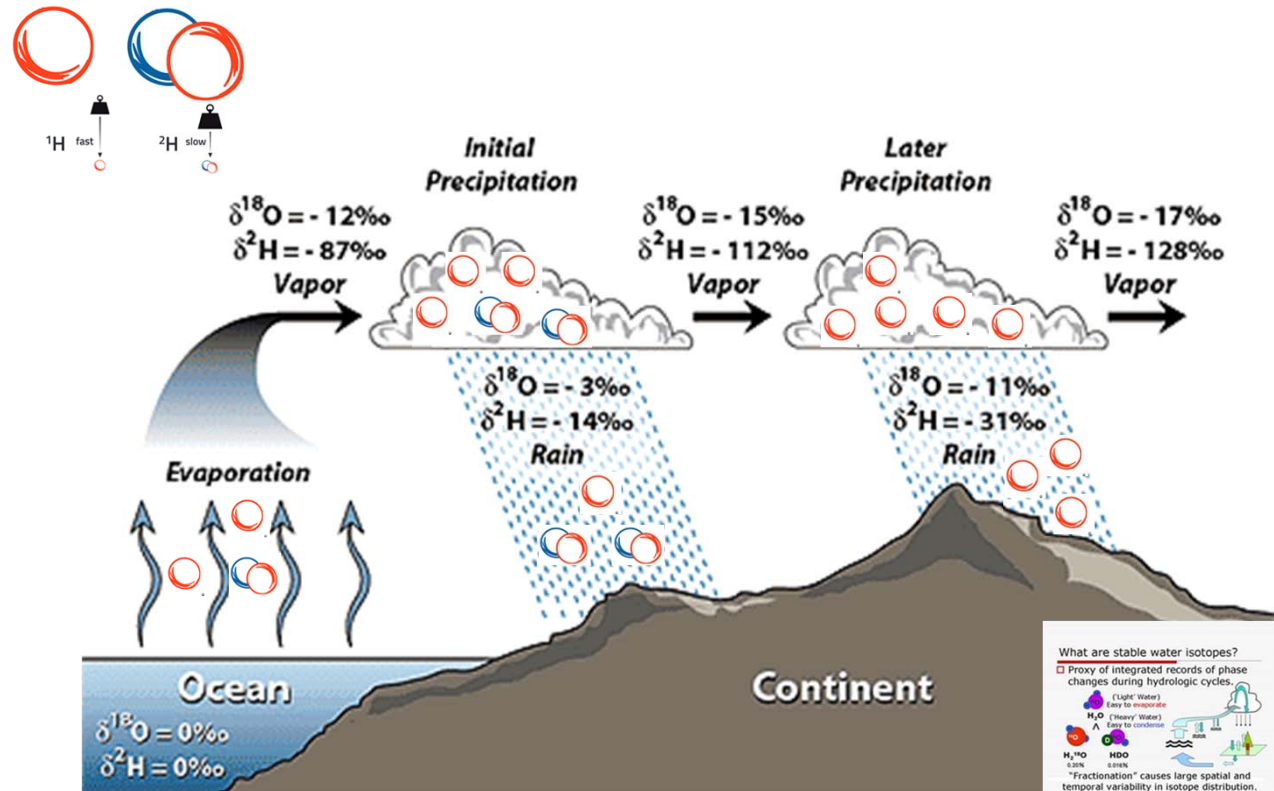
Fingerprinting & The Water Cycle

-Marine sources of water are fairly uniform and measure 0‰

-Vapor will favor ^1H and ^{16}O depleting the delta values.

-Precipitation is generally depleted but condensation will favor ^2H and ^{18}O enriching the delta values.

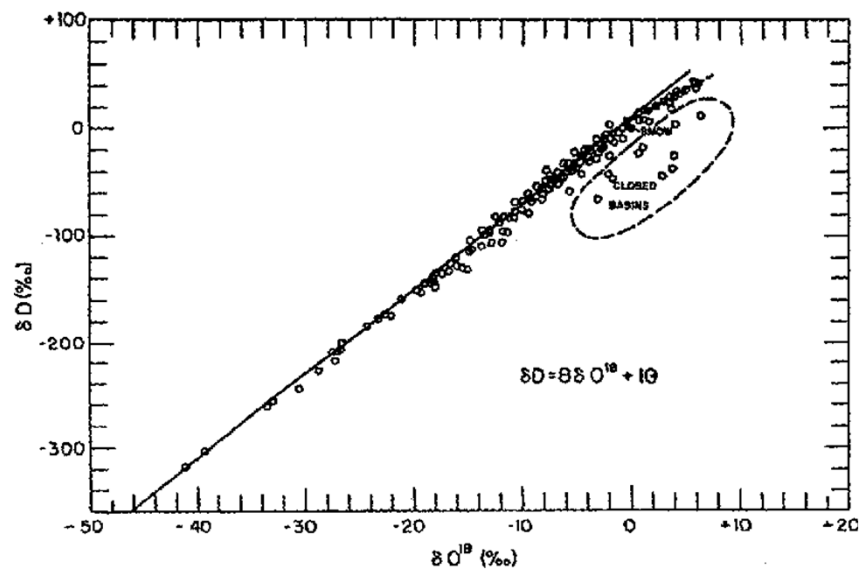
-Ground water from snow melt in high altitudes and polar regions will show significant depletion in delta values.



What are stable water isotopes?
 □ Proxy of integrated records of phase changes during hydrologic cycles.
 (Light Water) Easy to evaporate
 H_2O (Heavy Water) Easy to condense
 H_2^{18}O HDO
 0.20% 0.036%
 "Fractionation" causes large spatial and temporal variability in isotope distribution.

The Global Meteoric Water Line

-In 1961 Dr. Harmon Craig introduced the concept of the Global Meteoric Water Line which describes the relationship of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ isotopes of natural waters.



$$\delta\text{D} = 8 * \delta^{18}\text{O} + 10$$



Performing shipboard titrations on the 1961 MONSOON expedition to the Antarctic Ocean.

The Geochemical News #98, January 1999

Craig, H. (1961). Isotopic variations in meteoric waters. *Science*, 133(3465), 1702-1703.

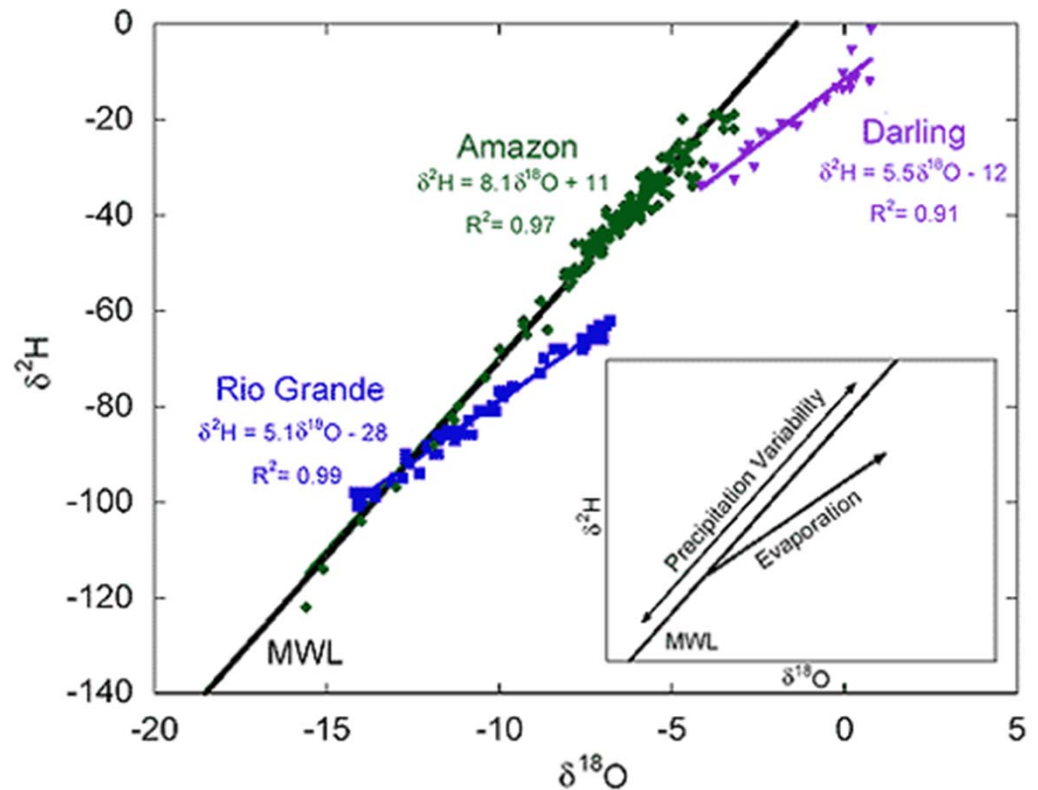
Fingerprinting & The Water Cycle

-The Global Meteoric Water Line

-The GMWL describe the relationship (slope) between $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values of water samples from around the globe and is considered the general trend that is expected within the water cycle.

-This relationship is key in utilizing isotopic data to source water.

-A Local Meteoric Water Line (LMWL) describe this relationship on a regional scale.

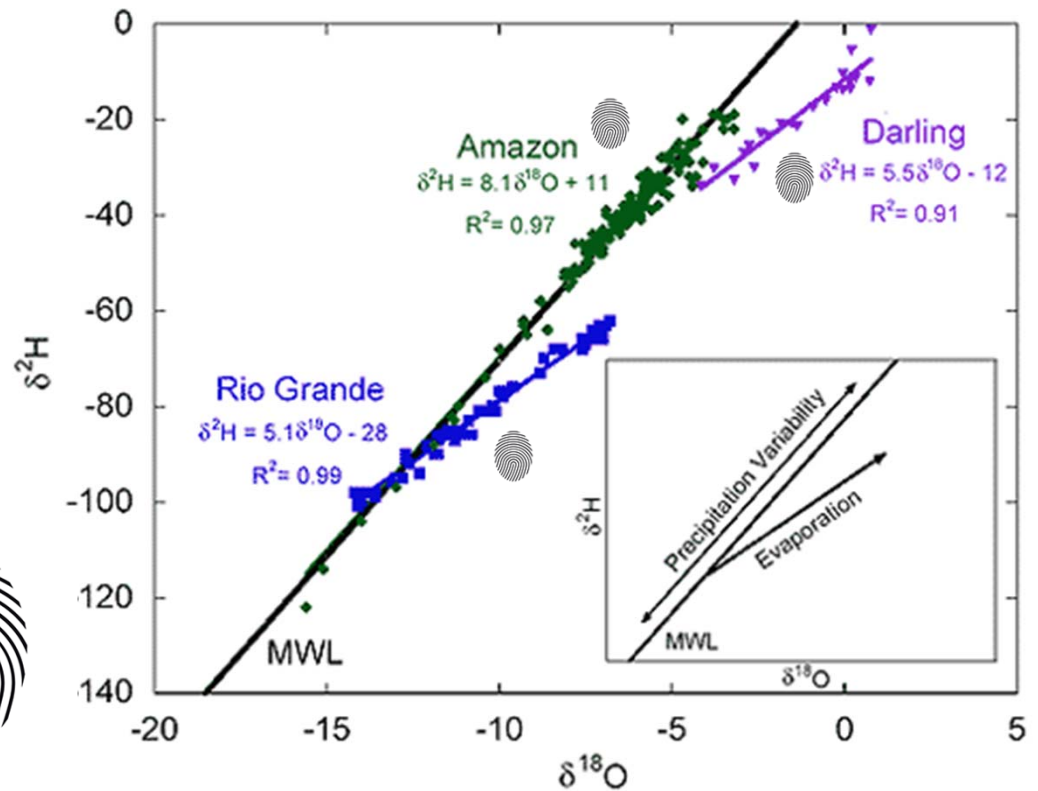
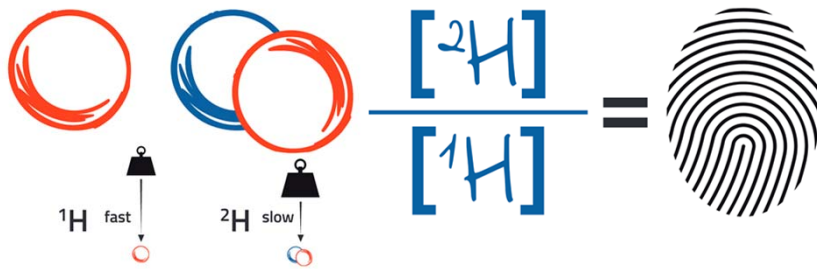


<http://web.sahra.arizona.edu/programs/isotopes/oxygen.html>

Fingerprinting & The Water Cycle

-Since it has been discovered that waters from different geographic regions have natural and distinct isotopic ratios, these ratios have been used to identify water sources.

-Utilizing the isotopic value and their covariance as a tracer or fingerprint has since become a very standard application of water isotopes.



<http://web.sahra.arizona.edu/programs/isotopes/oxygen.html>

$\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ Reveal the Sources of Nitrate-Nitrogen in Urban Residential Stormwater Runoff

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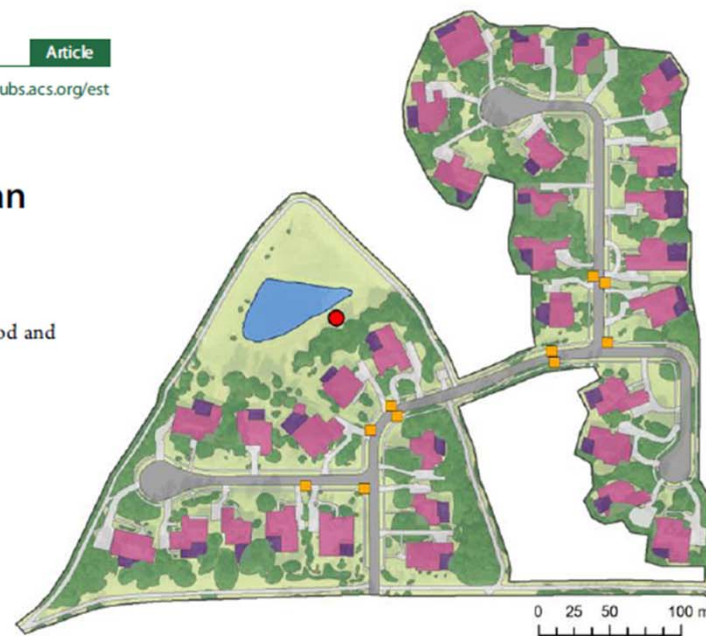
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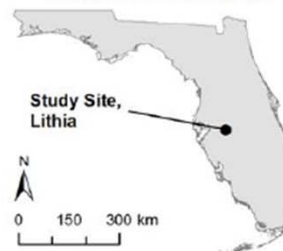
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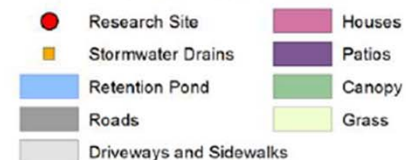
Nitrate (NO_3^-) $\rightarrow \delta^{18}\text{O}_{\text{NO}_3^-}$ and $\delta^{15}\text{N}_{\text{NO}_3^-}$



State of Florida



Map Legend



Water Isotope Fingerprint of Local Stormwaters

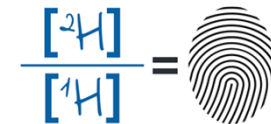
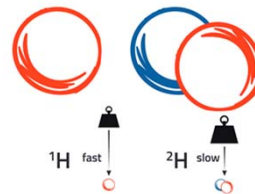
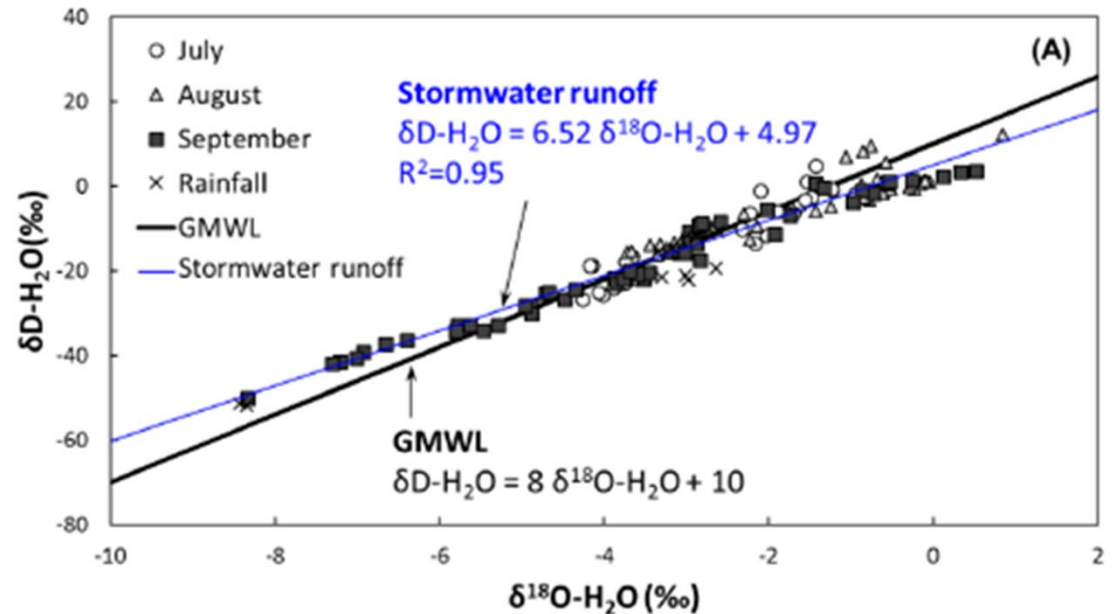
-Stormwater runoff during **all 25 storm events**

-Overall, our data indicated that all runoff water during 25 **stormwater events originated from the local rainfall and evaporation slightly changed the isotopic composition.**

-This suggests that no other sources of water (e.g., groundwater, municipal water, reclaimed water, leaking sanitary sewers) contributed any water and thus N in our stormwater runoff samples.

-Water isotopes were used as a diagnostic tool to measure the contribution of evaporative moisture. The mean d-excess values were lower than that GMWL indicated enrichment due to evaporation.

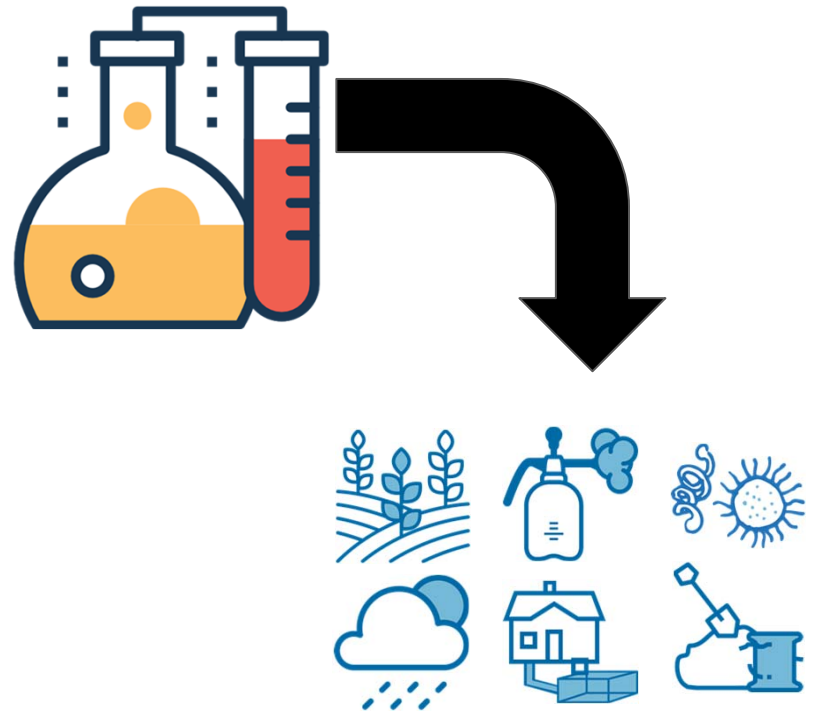
-Evaporation and impermeable areas leads to enrichment of the $\delta^2\text{H}$ values



Nutrient Source Tracking (NST)

Nutrient Source Tracking

- Issues with excess nitrate Point Source vs. Nonpoint Source
- Nitrate(NO_3^-) $\delta^{18}\text{O}$ and $\delta^{15}\text{N}$
- Nitrate isotopic fingerprinting
- Stormwater Case Study

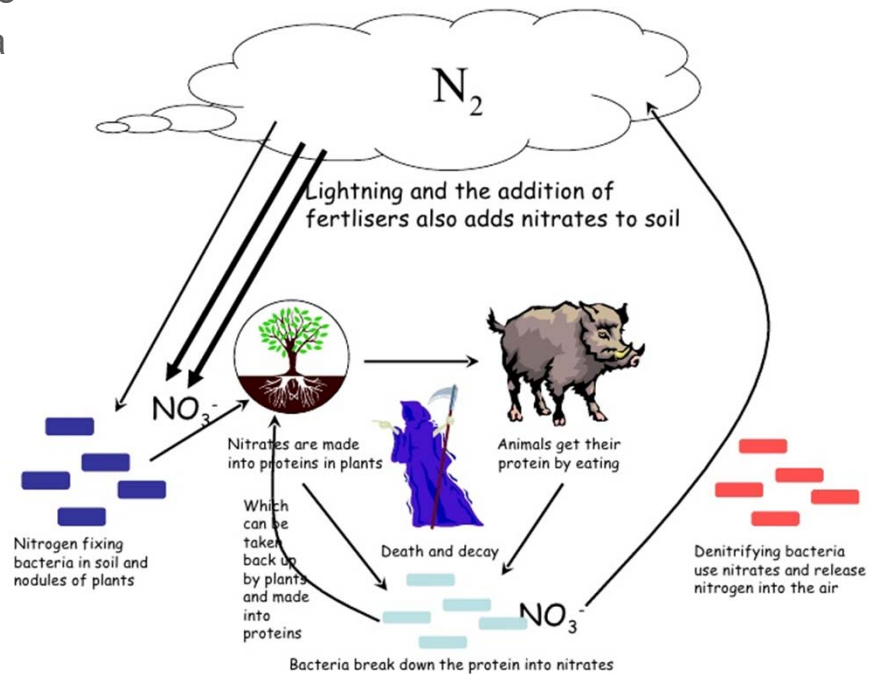
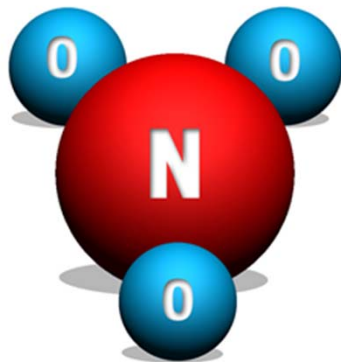


What is Nitrate?

-Nitrate, NO_3^- , is a naturally occurring molecule which contains 3 Oxygen atoms surrounding a central Nitrogen atom. In a trigonal planar structure.

-It is highly soluble in water

-Essentially a vehicle for nitrogen



Source: Dr. Shabeel Pn

URL:<https://www.slideshare.net/shabeelpn/nitrogen-cycle-3614281>

Long Term Environmental Hazards

-Eutrophication can greatly impact recreational water quality, deterring tourism.



-With eutrophication also comes **algal toxins**, marine algae produce toxins and cause large fish die offs. These poisons bioaccumulate and can eventually hit your dinner plate.

-Marine and Fishers industries struggle and due to a **dropping fish population**.



-All of which lead to **losses in revenue** for several industries which rely on

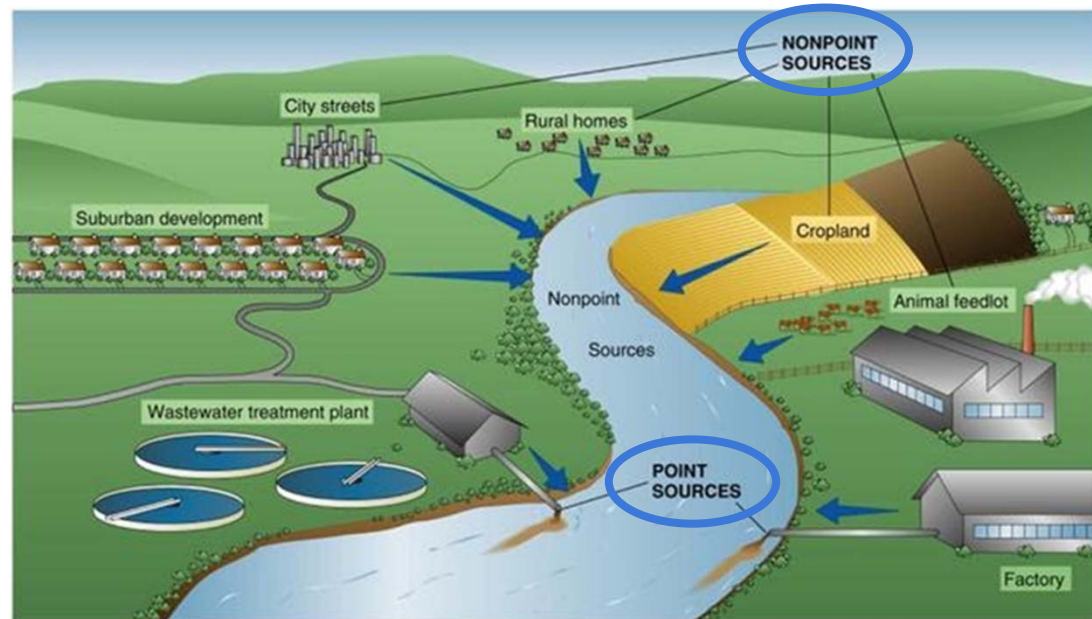


Nutrient Source Tracking

-Nonpoint source and point source contamination can be difficult to distinguish.

-Aside from obvious point sources such as wastewater treatment plants there are also less obvious groundwater **nonpoint sources** which can be difficult to account for.

-When mitigating nutrients or developing wetland projects it is necessary to understand the sources of nutrients into and out of a system.

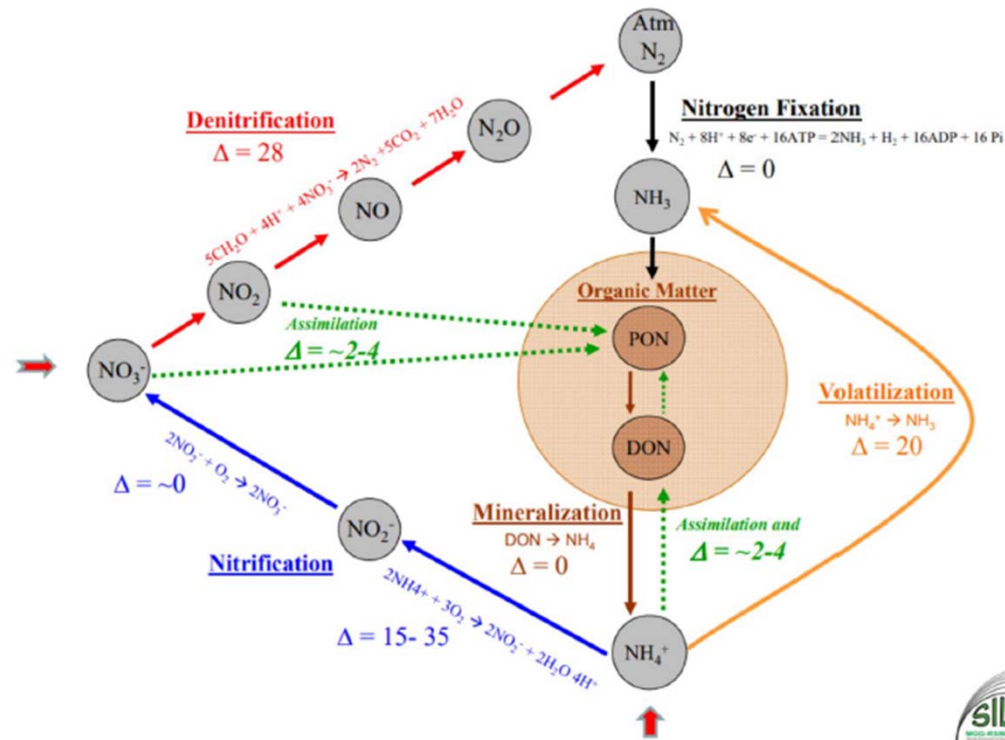


Isotopic Tracers Why Nitrate?

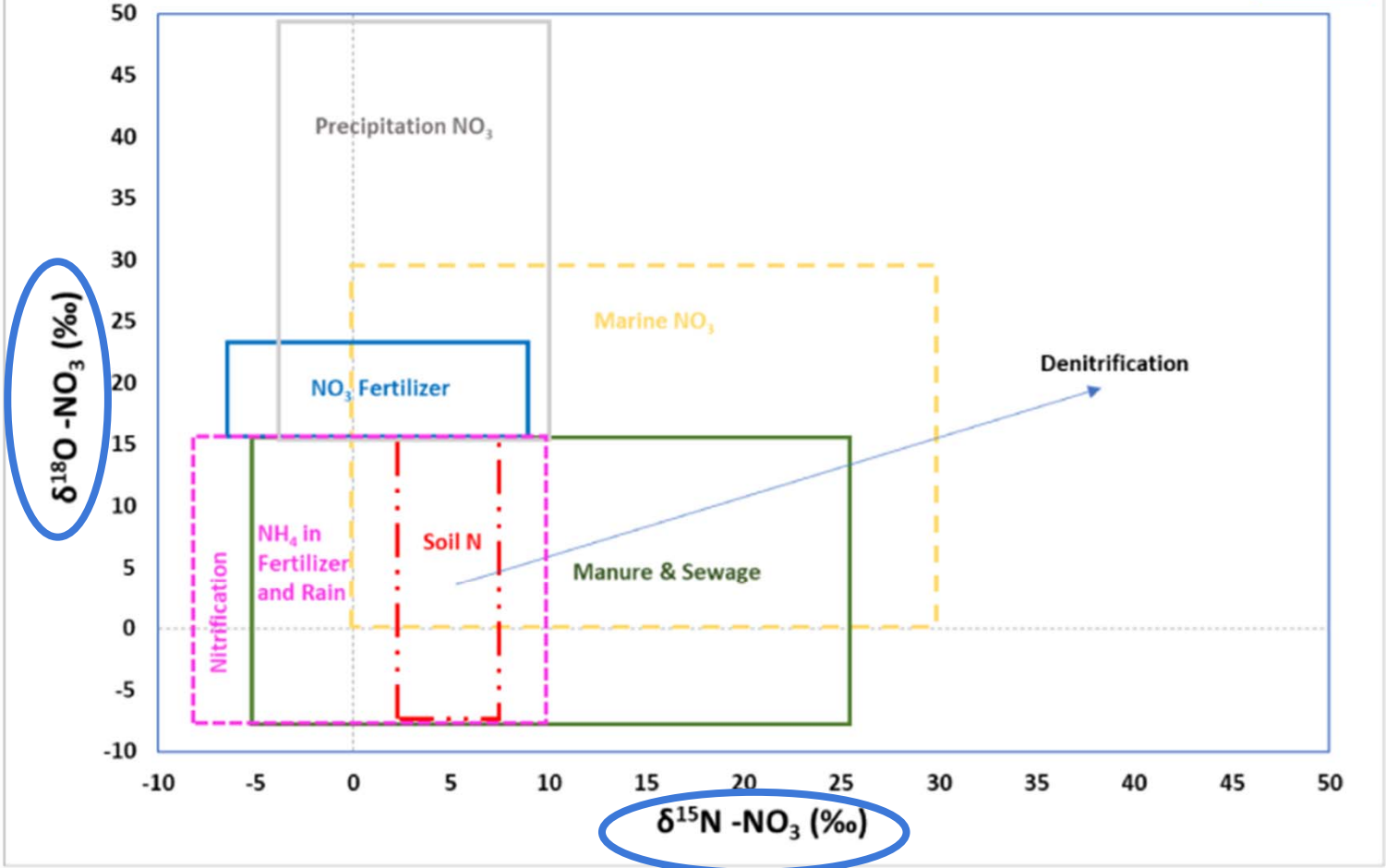
-Nitrate analysis allows for the **clearest understanding of fate** of nitrogen by giving insight into what pathway nitrogen is following.

-By having both oxygen and nitrogen isotopic values a **dual isotopic approach** can be made, similar to water, there is more evidence than compared to using one isotopes alone.

-Whether being nitrified or denitrified the **fractionation imposed on both isotopes is evident.**

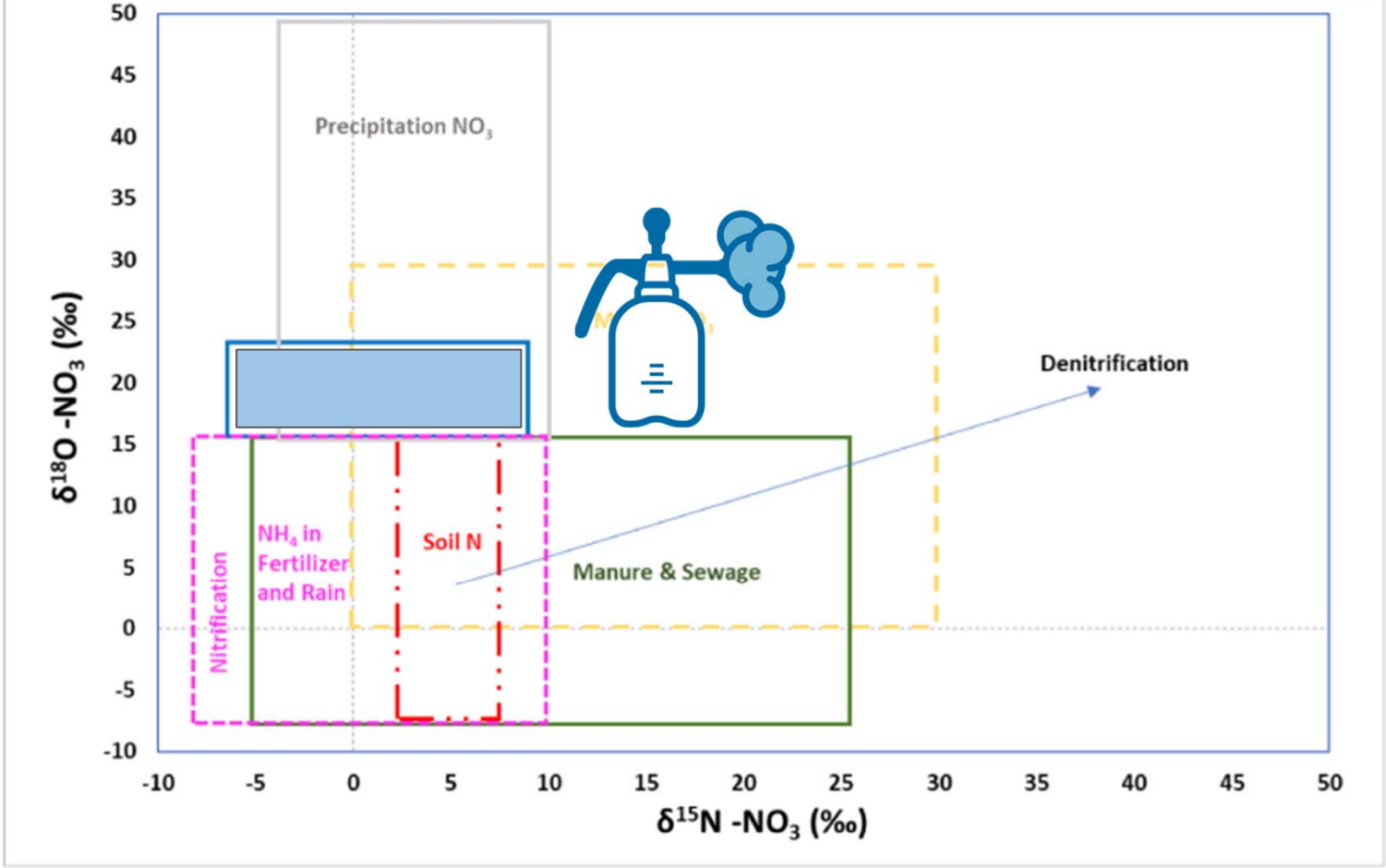


Potential Sources of Nitrate

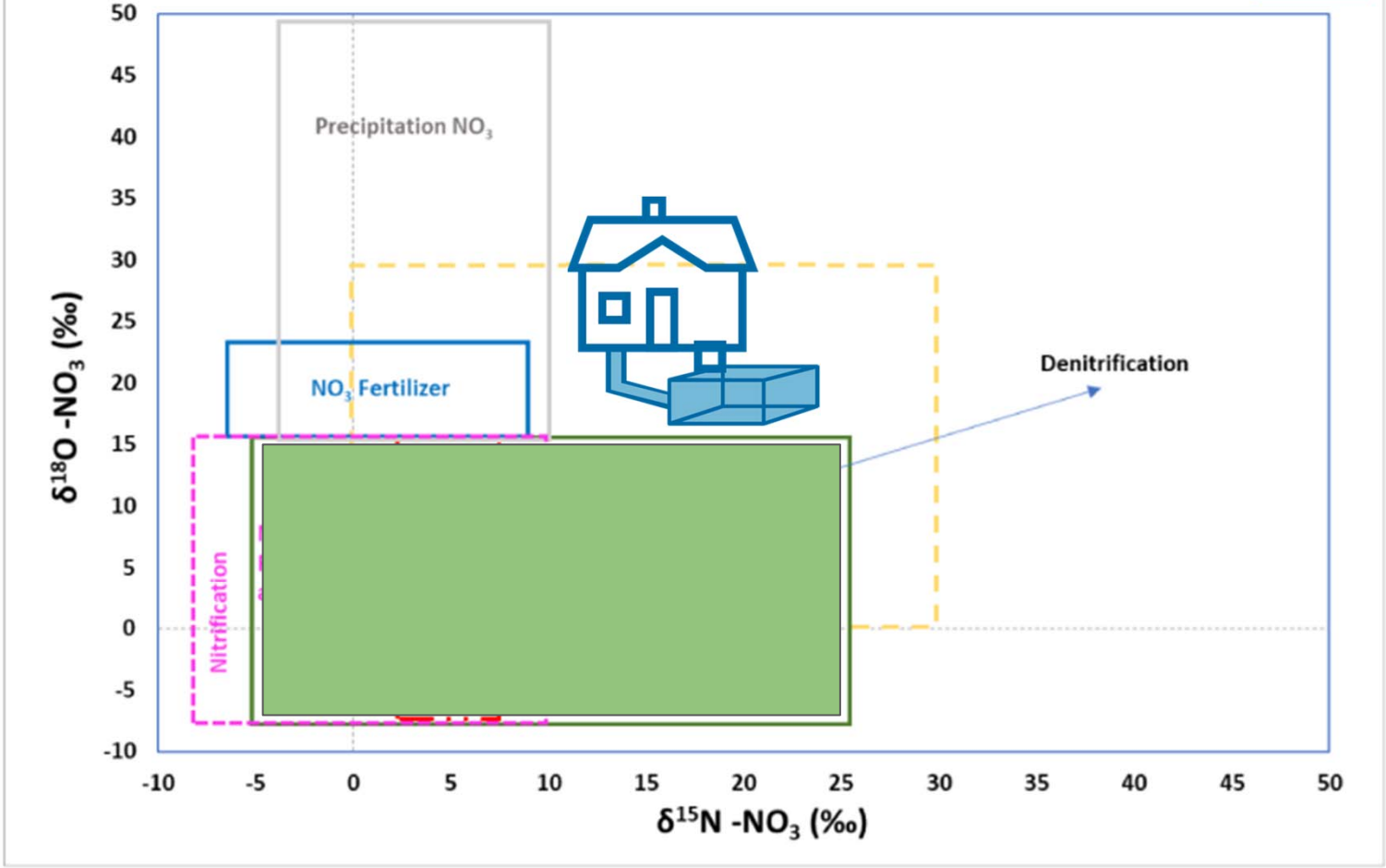


$\delta^{15}\text{N}-\text{NO}_3$ (‰)

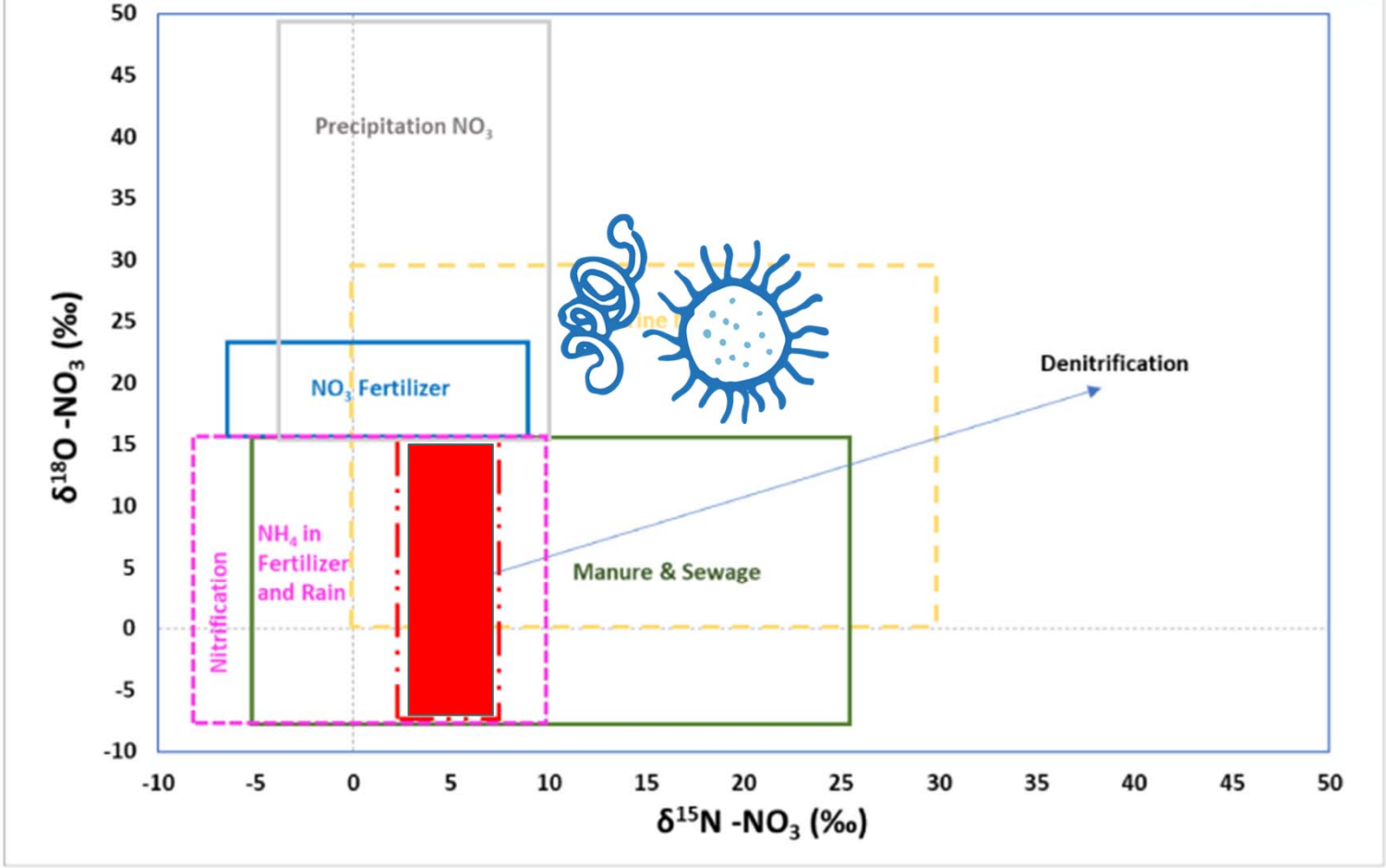
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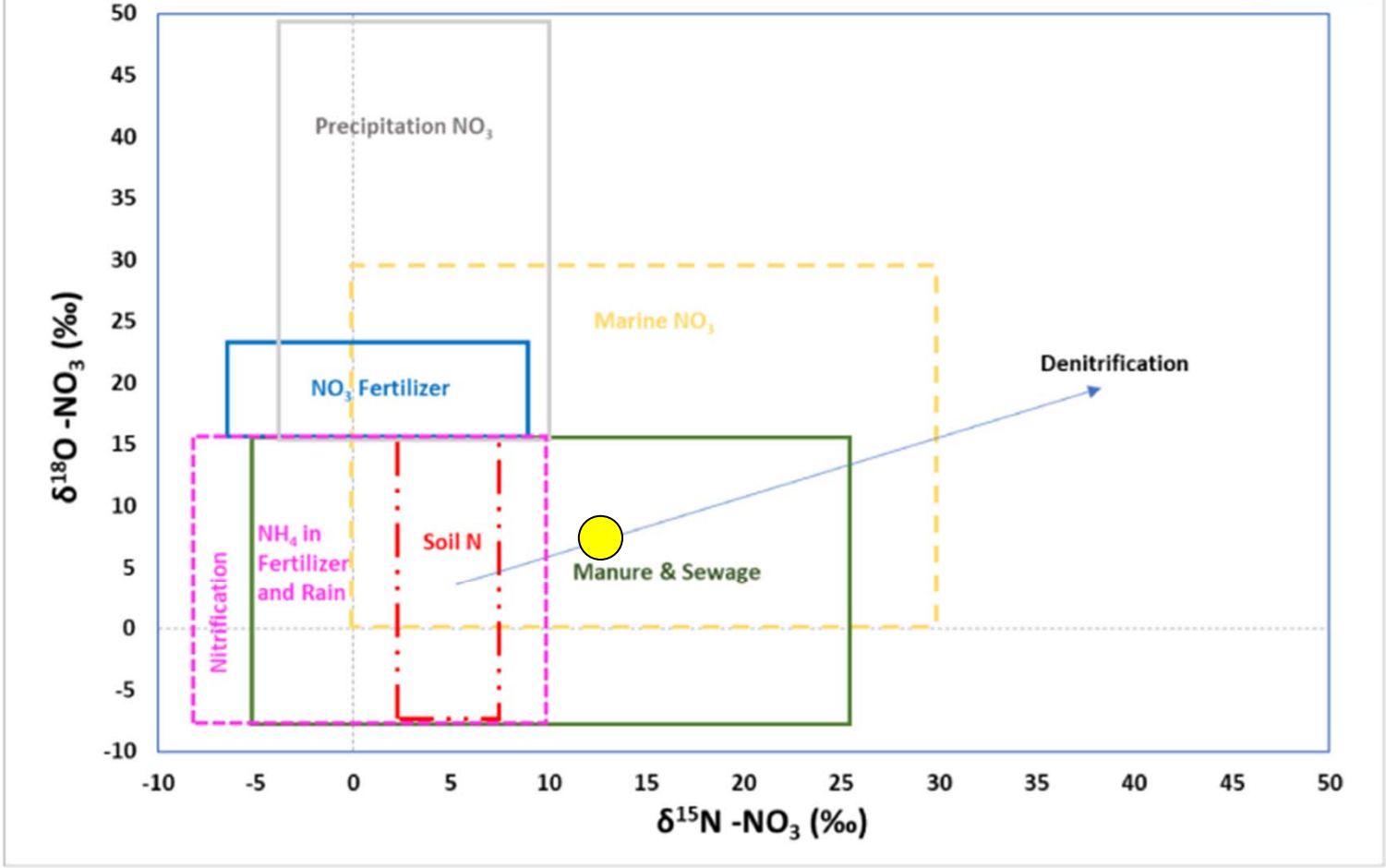
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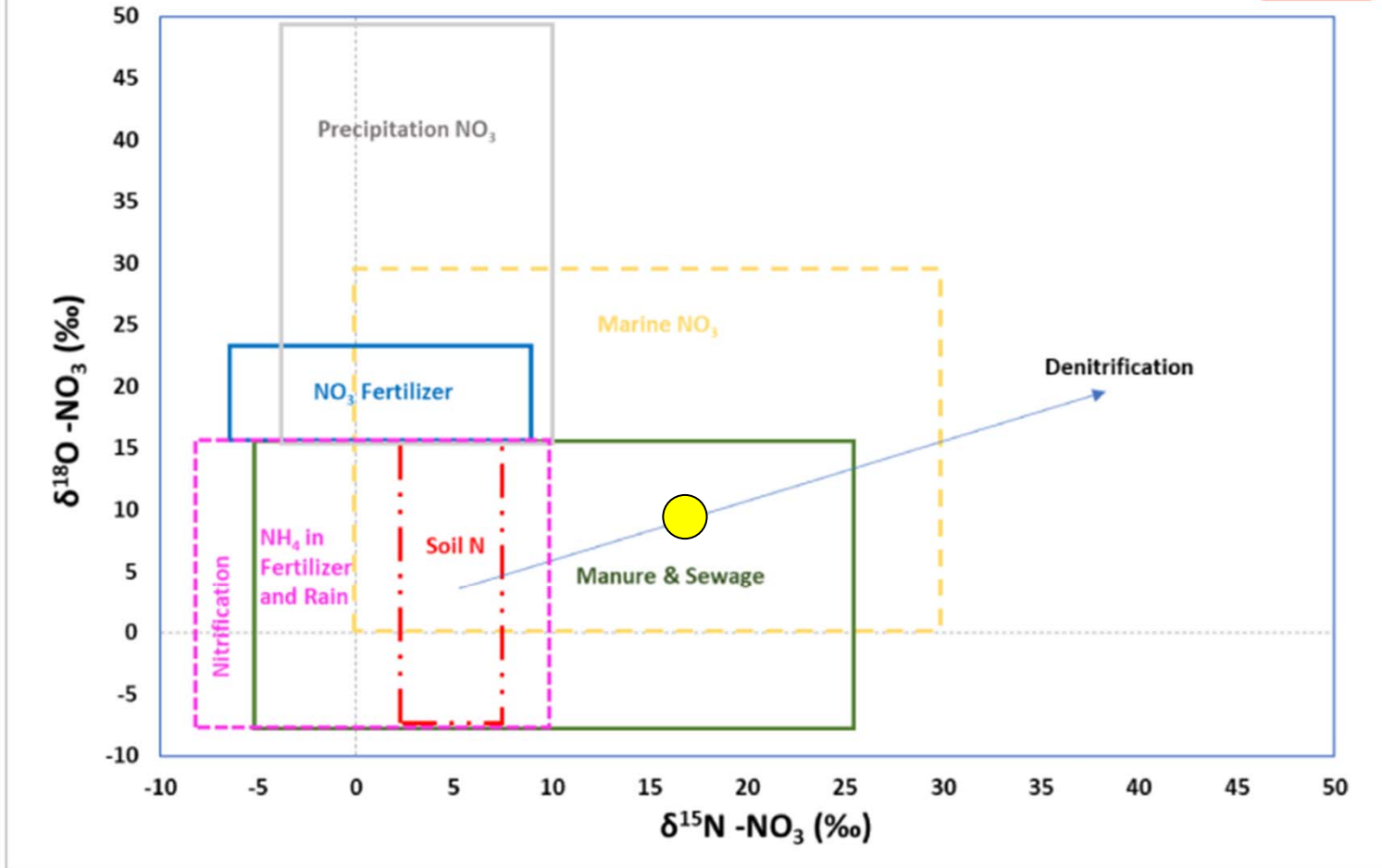
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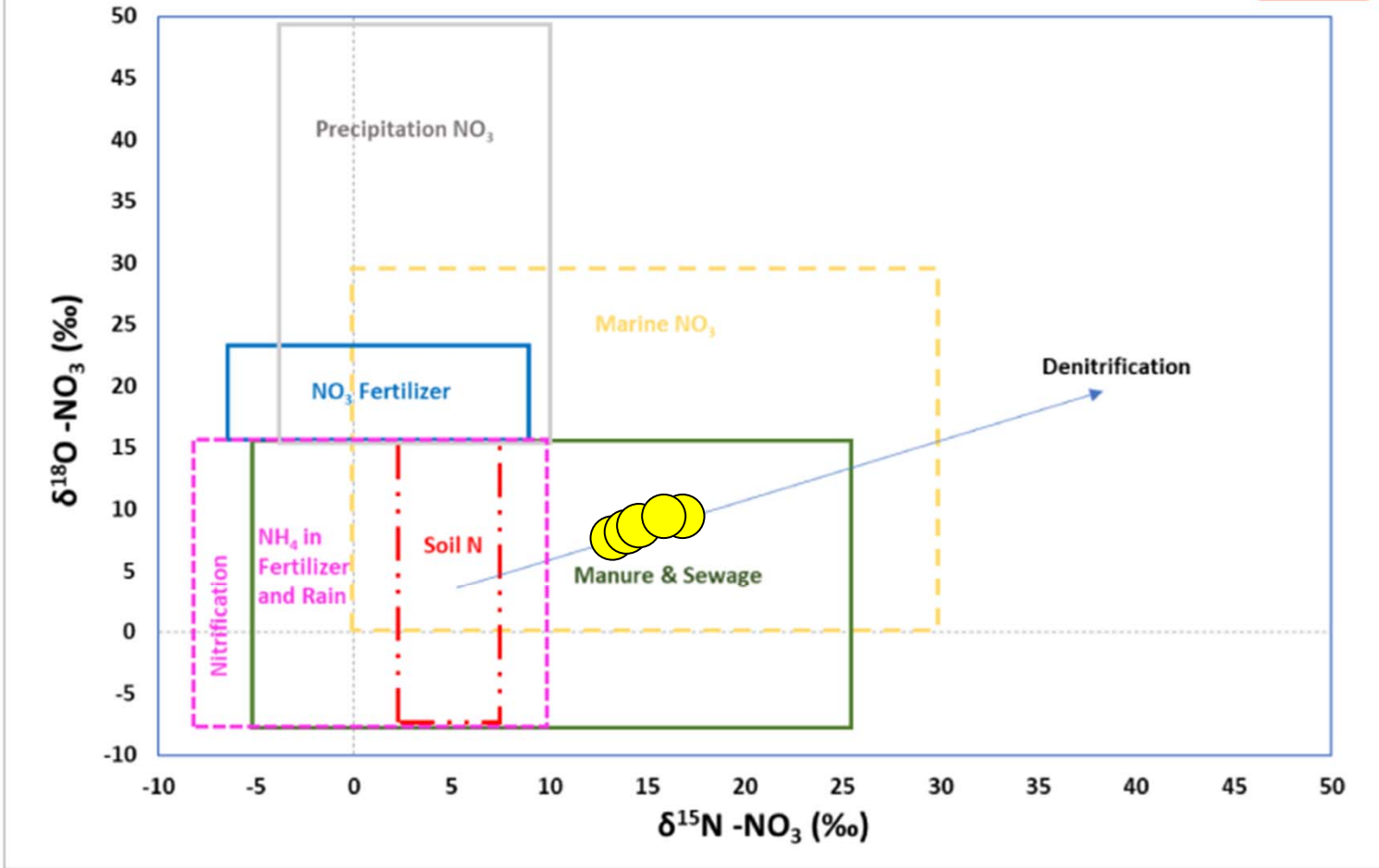
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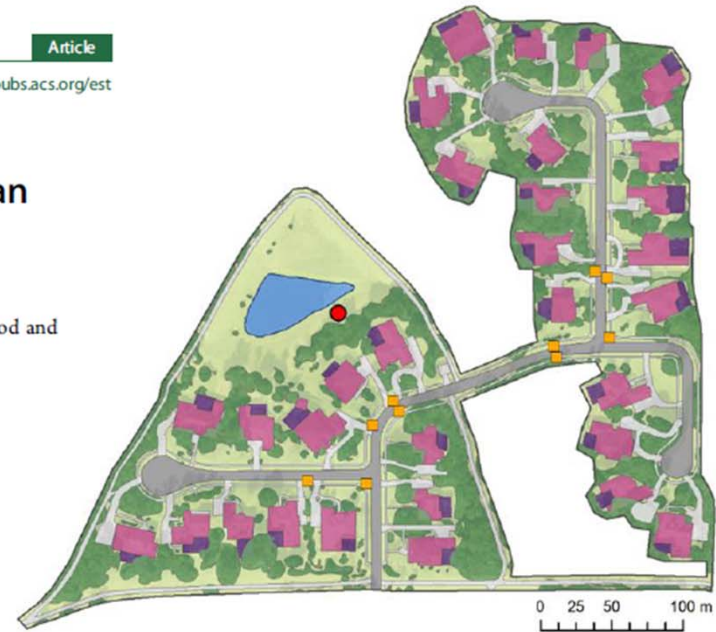
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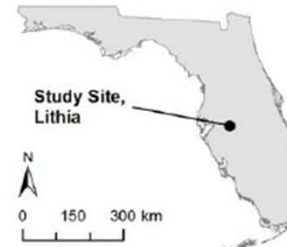
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State of Florida



Map Legend



Nitrate Isotope Dynamics

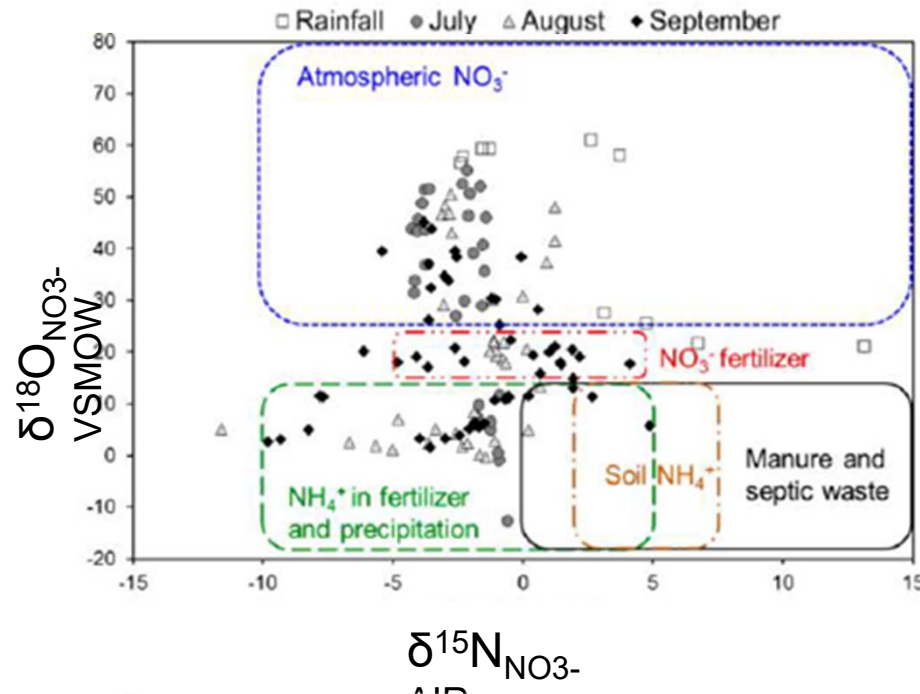
Nitrate Isotopic Results

Nitrate (NO_3^-) \rightarrow $\delta^{18}\text{O}_{\text{NO}_3^-}$ values varied from ~ 0 to $+60\text{‰}$

Nitrate (NO_3^-) \rightarrow $\delta^{15}\text{N}_{\text{NO}_3^-}$ values varied from ~ -10 to $\sim +5\text{‰}$

Rainfall results were on average enriched in $\delta^{18}\text{O}$ consistent with an atmospheric source.

Seasonally in July and August $\delta^{18}\text{O}$ results are on average more enriched than results in September.



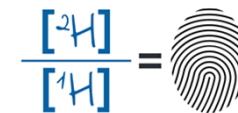
ENVIRONMENTAL
Science & Technology

Article
pubs.acs.org/est

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Nitrate Isotope Dynamics

Potential Sources of Nitrate

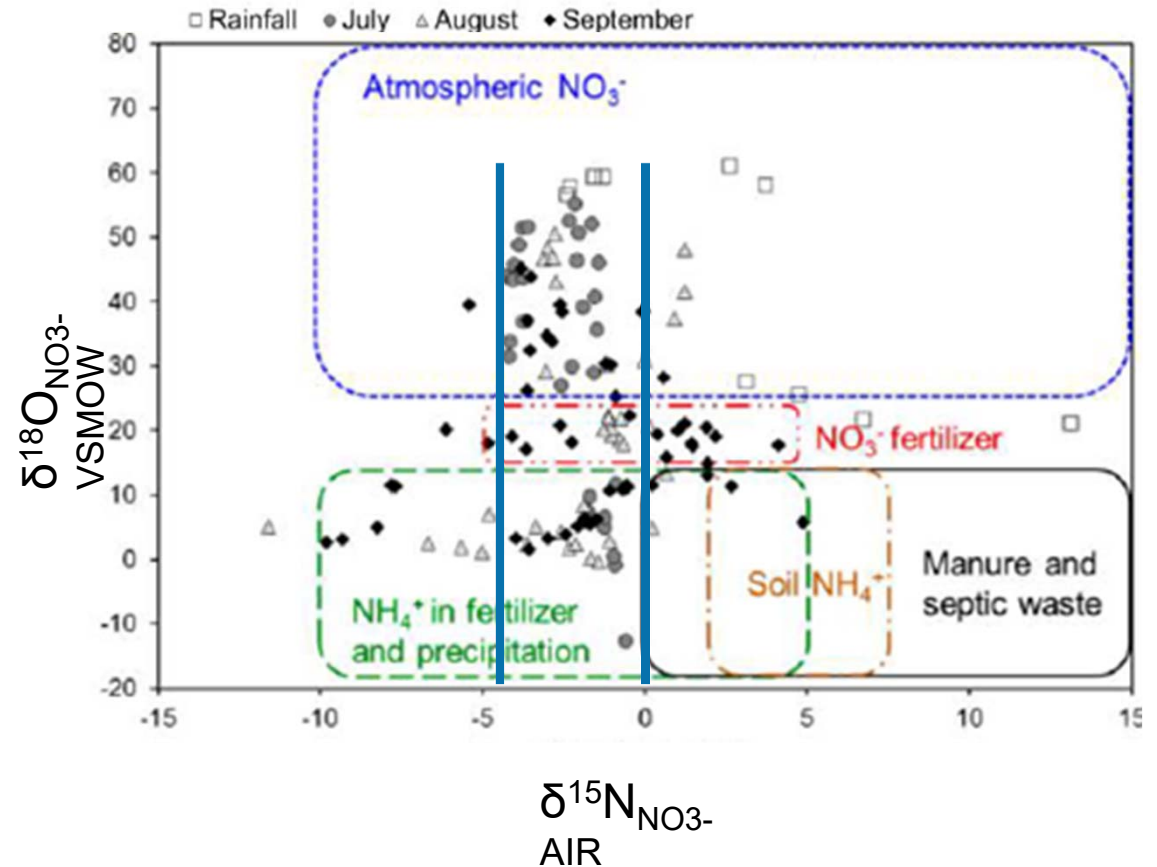
~60% of the runoff samples $\delta^{15}\text{N}_{\text{NO}_3^-}$ values varied from **0 to -4‰**

-The potential sources diagram suggests:

- Atmospheric deposition
- Chemical fertilizer
- Soil based N and
- Organic N

-Sources as the main contributors of NO_3^-

-No evidence of Manure or Septic waste.

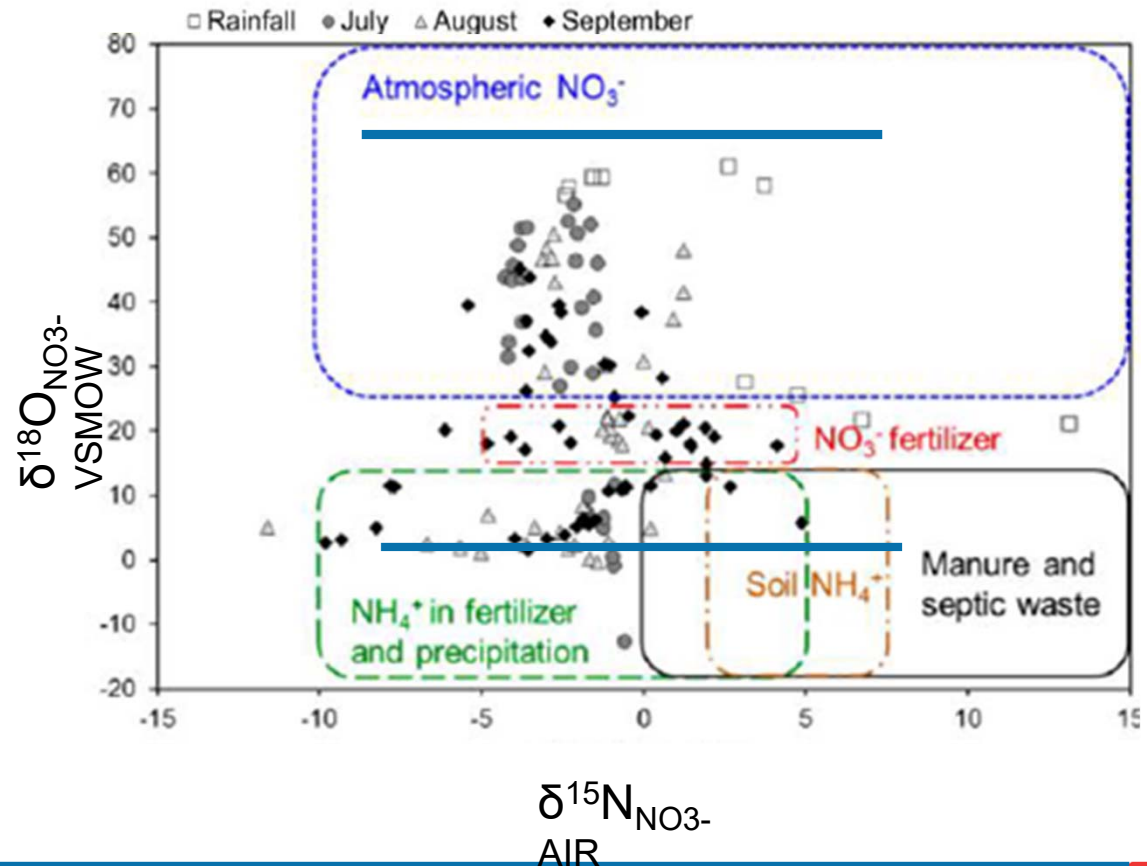


Nitrate Isotope Dynamics

Potential Sources of Nitrate

-Positive $\delta^{18}\text{O}_{\text{NO}_3^-}$ values suggest **atmospheric deposition** as a major contributor of nitrate to stormwater.

- $\delta^{18}\text{O}_{\text{NO}_3^-}$ values are known to be more useful tools when $\delta^{15}\text{N}_{\text{NO}_3^-}$ have wide ranges.



Nitrate Isotope Dynamics

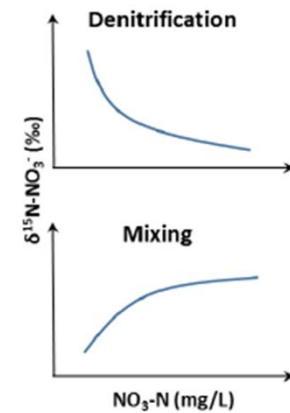
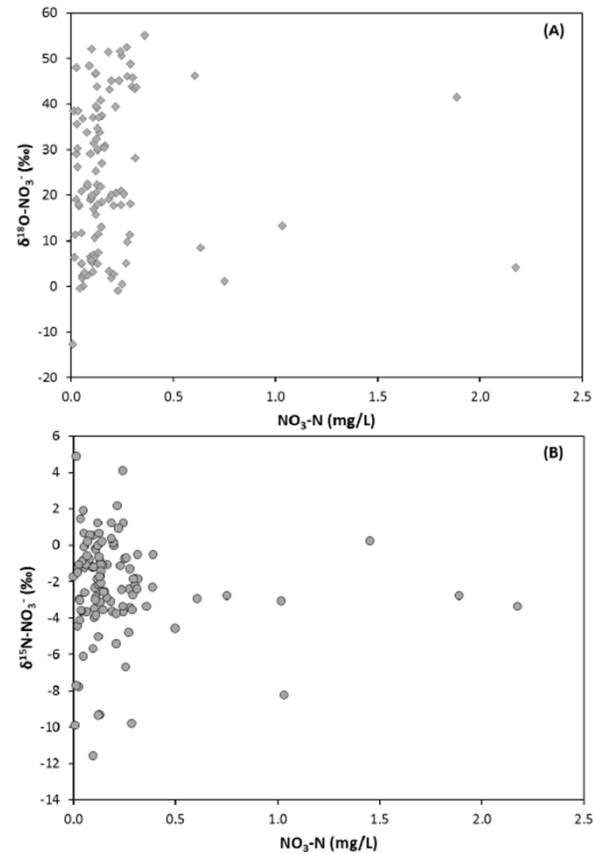
Mixing and Fate

-By combining isotopic data with concentration data, **evidence for denitrification (transport out the system) and mixing (transport into the system)** can be assessed.

-Denitrification (DN) is expected to increase delta values as concentrations decrease

-Mixing tends to increase both delta values and concentrations

-There was some evidence for mixing and no evidence for DN



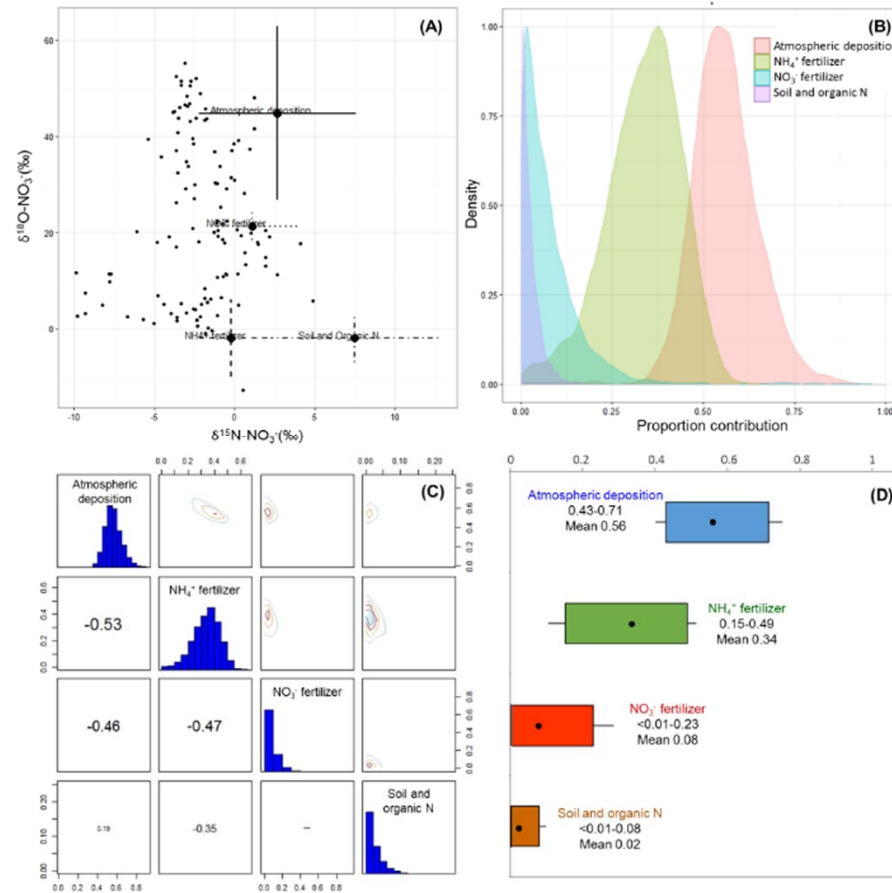
Nitrate Isotope Dynamics

Delineated Sources

-Statistical analysis was applied to the data sets to best determine sources in urban runoff.

-**Atmospheric deposition** was found to contribute more than half of the nitrate found in urban runoff at **~56%**.

-**Ammonia based fertilizer** was found to be the 2nd dominant source in urban runoff at **~34%**.



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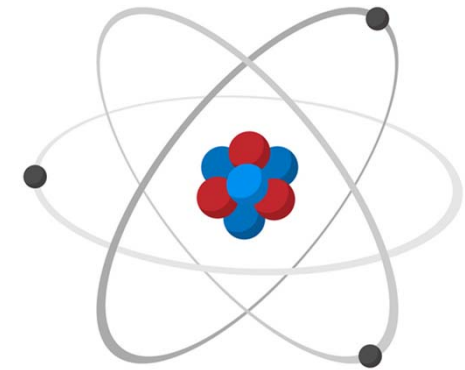
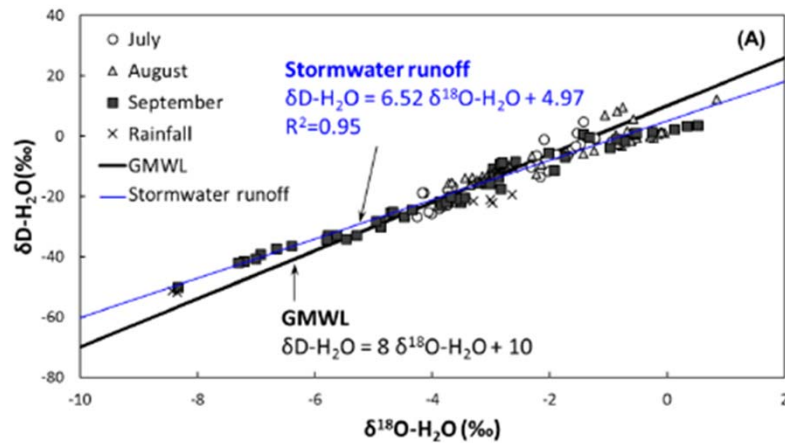
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Isotopic Applications Review

$\delta^{18}\text{O}_{\text{H}_2\text{O}}$ & $\delta^2\text{H}_{\text{H}_2\text{O}}$

-Interpretation of water sources, mixing, and source heterogeneity.

sowhat

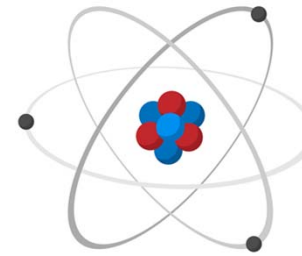
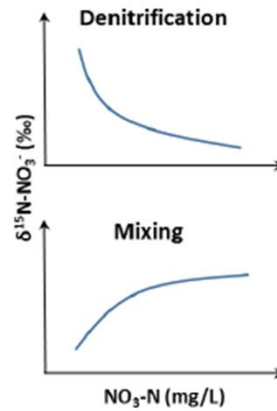
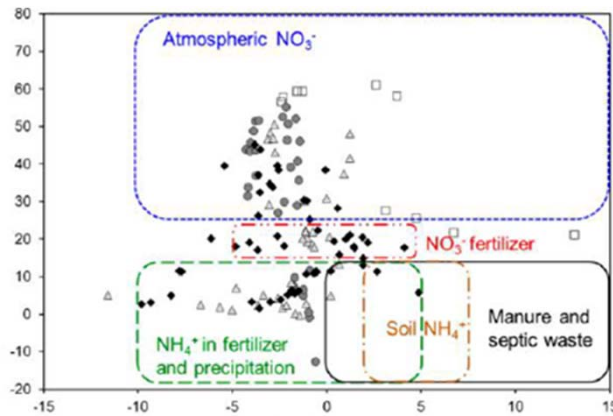


Isotopic Applications Review

$\delta^{18}\text{O}_{\text{NO}_3^-}$ & $\delta^{15}\text{N}_{\text{NO}_3^-}$

-Insight into source, transport mechanisms

sowhat



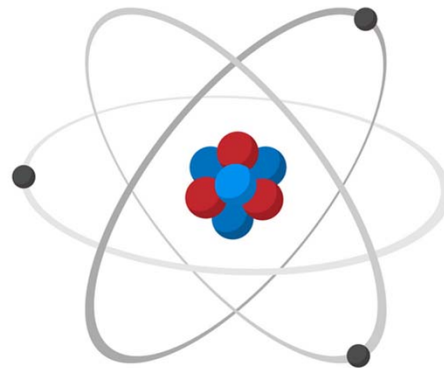
Isotopic Applications Review

-Insight provided by isotopic analysis into the source and transport of excess nutrients into stormwater system can help **optimize stormwater catchment areas.**

- This data can also be used to **suggest whether NPS or PS contamination** are the causes of elevated nutrients which may ultimately be introduced in groundwaters systems or directly into canals and estuaries.

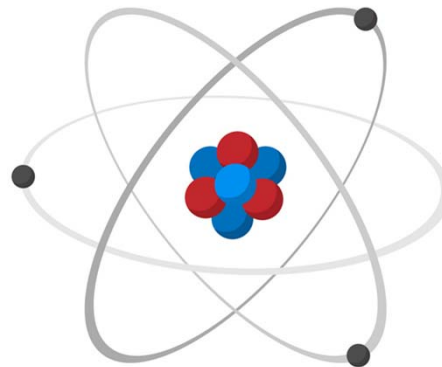
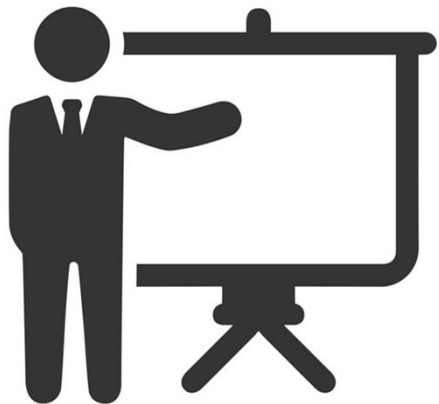
-All without the need for dosing or subsurface **intervention.**

sowhat



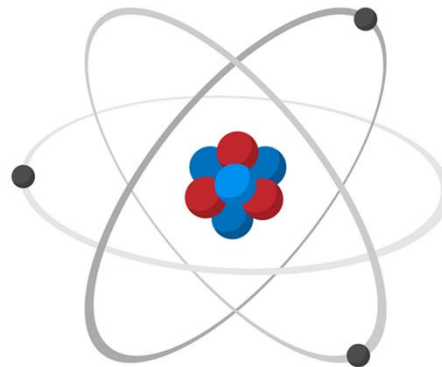
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- Explain how fingerprinting is used to ID and track water and nutrient sources.
- Review case study applications



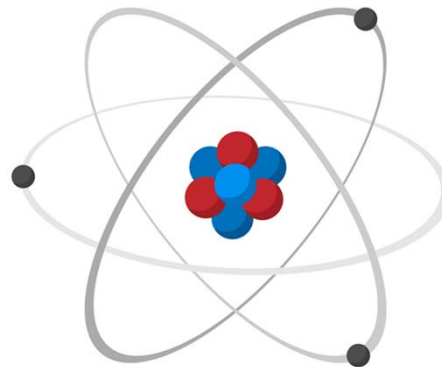
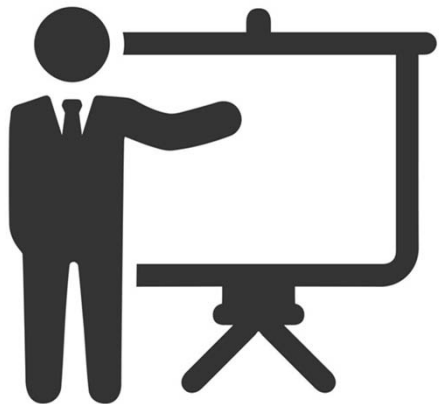
The Goals of Our Next 30 mins

- Define isotopic fingerprinting of water and nutrients.
- Explain how fingerprinting is used to ID and track water and nutrient sources.
- Review case study applications



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Summary and Review

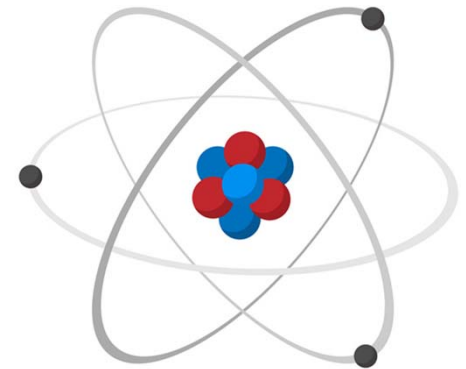
Isotopic Tracers have allowed for:

-Insight into the source and transport of excess nutrients into stormwater system can help **optimize stormwater catchment areas** as well be used to suggest whether NPS or PS contamination are the causes of elevated nutrients which may ultimately be introduced in groundwaters systems or directly into canals and estuaries.

-These finding suggested **the importance of atmospheric contributions in the area, as well as mixing with fertilizers.**

- There was **evidence that fertilizer nitrification was promoting the formation of NO_3^{2-}** . With no evidence for denitrification.

So what





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Questions?

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